Glutaminase-γ-Glutamyltransferase: Subcellular Localization and Ammonia Production in Acidosis (40335)

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In the rat kidney, glutamine is utilized by either the mitochondrial glutaminase 1-glutamate dehydrogenase pathway (1-3) or a glutaminase- γ -glutamyltransferase pathway (4-6). The subcellular location of the glutaminase- γ -glutamyltransferase is unknown, although if localized in the cytosol, it would lend support to the previously proposed hypothesis (7) postulating glutamine utilization by dual pathways. The present study was designed (i) to determine its subcellular location and (ii) to determine its quantitative contribution to ammonia production by acidotic kidneys perfused with 1 mM L-glutamine.

Materials and Methods. Male Sprague-Dawley rats, weighing between 350 and 400 g, were tube fed 1400 μ moles of NH₄Cl (0.40 *M*) per day for 3 days; at the end of the second day they were placed in metabolic cages, one per cage, and 24-hr urine collections were observed. Throughout the study the animals were maintained on rat chow (Purina) and water *ad libitum*. A constant ingestion of NH₄Cl ensured a similar degree of acidosis in all rats; this was confirmed by monitoring systemic blood pH and HCO₃ concentration (Radiometer pH-bloodgas analyzer) at time of perfusion and determining 24-hr ammonium excretion.

Two hours prior to perfusion, rats were injected with either methionine-DL-sulfoximine (Sigma), 1.8 mmoles kg⁻¹, ip, dissolved in 1.0 ml of 0.9% saline or 0.9% saline alone. The animals were anesthesized with sodium pentobarbital, 30 mg kg⁻¹, ip, and their kidneys were isolated (8, 9) and perfused with an artificial plasma solution containing albumin (Sigma, fraction IV) and 1 mM Lglutamine; albumin was defatted (10) and dialyzed, two changes, for 48 hr against 4 liters of the perfusate solution, minus albumin. Kidneys were perfused, pH 7.40, with 80 ml of perfusate for 60 min and samples of the perfusate media were taken at 15-min intervals. The media were analyzed for ammonia concentration by both the enzymatic (6) and the Conway microdiffusion methods, modified for blood ammonia (4); glutamine concentration was determined by measuring liberated ammonia after enzymatic (*Escherichia coli* glutaminase, Sigma) deamidation (3, 9). Ammonia production and glutamine uptake rates were calculated as described (3, 9).

Following perfusion, acidotic and acidotic plus MSO-treated rat kidneys were homogenized in ice-cold 0.44 M sucrose containing 50 mM MgCl₂ and 2 mM HEPES, pH 7.4. Subcellular fractionation was carried out according to a standard schedule (11) on a Sorval RC2B refrigerated, 0-4°, centrifuge; the postmitochondrial fraction was transferred to a Beckman L ultracentrifuge and centrifuged at 105,000 g for 1 hr. The fractions obtained, nuclear + cellular debris, mitochondrial, microsomal, and soluble, were resuspended in fresh homogenizing solution and suitable aliquots were assayed for NH₃ and glutamohydroxamate formation by the γ -glutamyltransferase reaction (5). Protein content was determined using the biuret reaction (12) employing bovine albumin (Sigma, fraction IV) as the standard.

Results. The response to the standard NH₄Cl load is shown in Table I. Both groups, control and pre-MSO-treated rats, received an identical acid load, exhibited a similar degree of mild acidosis, and excreted identical amounts of ammonium (coefficient of variation, 6.2 for control and 4.5 for pre-MSO-treated rats). Differences in ammonia production by perfused kidney from MSO-injected rats are not, therefore, due to a variable response to the acid load.

The effect of MSO on ammonia release and glutamine uptake is presented in Table II. Kidneys released 50 ± 4 and $48 \pm 6 \mu$ moles of ammonia g⁻¹ hr⁻¹ in the absence of exogenous glutamine. In the presence of gluta-

			Blood		
Pretreated		Intake ^a NH₄Cl (µmoles day ⁻¹)	pH (U)	HCO ₃ ⁻ (mEq li- ter ⁻¹)	Excreted NH ₄ ⁺ $(\mu moles$ day ⁻¹)
Group (4)	1 ^b	1400	$7.32^{\circ} \pm 0.06$	23.4 ±1.5	1358 ±85
Group	2	1400	7.34 ±0.07	22.8 ±0.9	1375 ±62

 TABLE I. Ammonium Chloride Intake, Systemic

 Acid-Base Balance, and Ammonium Excretion.

^{*a*} Given as 0.4 *M* NH₄Cl, 1400 μ moles day⁻¹ for 3 days (see Methods).

^b To be administered, 0.9% NaCl, 1 ml, ip.

^c Means \pm SEM from four rats.

^d To be administered, 0.9% NaCl plus MSO, 1.8 mmoles kg^{-1} .

 TABLE II. THE EFFECT OF MSO ADMINISTRATION ON AMMONIA RELEASE AND GLUTAMINE UPTAKE.

	Ammonia released	Glutamine uptake		
	$(\mu \text{moles } \mathbf{g}^{-1} \mathbf{h} \mathbf{r}^{-1})$		Ammonia/ glutamine	
Control ^a				
$0 \text{ m} M (4)^{b}$	50 ± 4			
1 mM(4)	241 ± 24	119 ± 13	2.02	
Δ^{c}	191 ± 18	119 ± 13	1.60 ^e	
MSO^{d}				
$0 \mathrm{m}M(4)$	48 ± 6	_	2.86	
1 mM(4)	157 ± 15	55 ± 9	1.98^{e}	
Δ ^c	109 ± 11^{e}	55 ± 9°		

^a Control acidotic rats (Table I).

^b Number of rats.

^c Ammonia released with 1 mM glutamine-0 mM glutamine.

^d MSO, 1.8 mmoles kg^{-1} given 2 hr prior to perfusion.

^e Significantly different from control (P < 0.05).

mine, control acidotic kidneys released 241 \pm 24 µmoles of ammonia while MSO-treated rats released significantly less (P < 0.01), 157 \pm 15 µmoles of ammonium. Glutamine uptake for acidotic controls was 119 ± 13 μ moles, which fell to only 55 ± 9 μ moles (P < 0.001). If one assumes ammonia released, in these studies, accurately reflects production, then the NH₃ produced per glutamine extracted ratios are 2.02 ± 0.05 for acidotic control, and 2.85 ± 0.08 for acidotic plus MSO-treated kidneys. Since a value of 2.0 is the highest possibly attained from complete deamidation and deamination of glutamine, it is clear that ammonia released in the absence of glutamine contributes to the total released in the presence of glutamine. Subtracting this gives an NH₃/Gln ratio of 1.60 in the acidotic control and 1.98 in the MSOtreated acidotic rats.

A direct effect of MSO on ammonia production from glutamine can be shown by adding the inhibitor to the perfusate (Fig. 1). Over the 30-min control period, the production rate averaged 1.43 µmoles min⁻¹; within 10 min production rates fell to 1.04 µmol min⁻¹. The fall in glutamine uptake was disproportionately greater than with ammonia production, falling from 25.9 \pm 3.2 µmoles per 30 min to 16.0 \pm 2.9 µmoles per min (*P* < 0.05). Consequently, the ammonia produced per glutamine extraction ratio rose from 1.66 to 1.95.

The subcellular localization of the glutamine-utilizing enzyme is shown in Table III. The activity, measured as both a glutaminase (ammonia liberated in absence of NH₂OH) and γ -glutamyltransferase, appears to be a soluble enzyme for the following reasons. The activity is mainly in the soluble fraction, 56% of the total homogenate activity, and its specific activity is significantly enriched only in



FIG. 1. Acute effect of methionine-DL-sulfoximine, 120 mg in 1.0 ml of 0.9% saline, on ammonia production from glutamine. Results are from four rats given the standard acid load (see Methods). Ammonia produced represents the total released minus the amount released in the absence of glutamine. Kidneys were perfused with 80 ml of perfusate; production rate is linear over the 60min period in the absence of MSO.

	Ammonia ^a		γ-GHA ^b	
Fraction	Total activity ^c	S . A . ^{<i>d</i>}	Total activity	S.A.
Homogenate	<u></u>			
Control	215 ± 16	1.5 ± 0.1	860 ± 45	6.0 ± 0.3
MSO	18 ± 5	0.1 ± 0.0	83 ± 8	0.6 ± 0.1
Nuclear				
Control	57 ± 8	1.8 ± 0.2	146 ± 13	4.6 ± 0.5
MSO	19 ± 11	0.6 ± 0.4	12 ± 3	0.4 ± 0.1
Mitochondrial				
Control	16 ± 5	0.3 ± 0.1	69 ± 14	1.4 ± 0.3
MSO	10 ± 7	0.2 ± 0.1	9 ± 3	0.2 ± 0.1
Microsomal				
Control	47 ± 6	2.2 ± 0.3	160 ± 21	7.4 ± 0.9
MSO	9 ± 4	0.4 ± 0.2	25 ± 12	1.0 ± 0.5
Soluble				
Control	98 ± 12	2.8 ± 0.3	481 ± 33	13.8 ± 1.0
MSO	6 ± 3	0.1 ± 0.1	19 ± 5	0.5 ± 0.1

TABLE III. RENAL GLUTAMINASE-7-GLUTAMYLTRANSFERASE: S	SUBCELLULAR	LOCALIZATION	AND N	ASO
Inhibition.				

^a Ammonia produced in the absence of NH₂OH.

^b γ-Glutamohydroxamate formed in the presence of NH₂OH.

^c Total activity, μ moles hr⁻¹, per fraction. ^d Specific activity, μ moles hr⁻¹, per mg of protein.

^e Mean \pm SEM from four kidneys in each group.

the soluble fraction. Both ammonia production and γ -GHA formation were markedly inhibited in the soluble fraction to values less than 15% of the control. Noteworthy ammonia production by the mitochondrial fraction (glutaminase 1 pathway) was unaffected.

Discussion. The results clearly demonstrate the inhibition of a glutaminase- γ -glutamyltransferase activity localized in the soluble fraction (Table III) which contributes 30 to 40% of ammonia produced by these mildly acidotic kidneys (Table II and Fig. 1). The disproportionately greater fall in glutamine uptake, 54%, than in ammonia production, 43%, is consistent with the glutaminase- γ -glutamyltransferase pathway contributing only one ammonia per glutamine. The rise in the NH₃ produced/glutamine extracted ratio to 2.0 after inhibition of the cytoplasmic pathway is consistent with complete deamidation and deamination by the mitochondrial pathway. These results therefore support the previous proposal of dual glutamine-utilizing pathways in the rat kidney with NH₃/Gln ratios reflective of the contribution from each pathway.

The present study underlines an important point in calculating the ammonia produced to glutamine extraction ratio (Table II). It must be realized that total ammonia release is not necessarily equivalent to that produced from the glutamine extracted. Thus, Hems (13) observed that nonacidotic kidneys perfused with 1 mM L-glutamine released 119 μ moles of NH₃ per 45 μ moles of glutamine, giving an NH₃/Gln ratio of 2.64; ammonia released in the absence of glutamine was similar to the present study, some 47 ± 4 μ moles. Since a ratio of greater than 2 is clearly impossible, subtracting the glutamineindependent release, 47 μ moles, from 119 gives 72 actually produced from glutamine and an NH₃/Gln ratio of 1.6. Ross (14) calculated an ammonia recovered to glutamine removed ratio of 1.9 with 1 mM L-glutamine as the substrate; however, if 47 μ moles of glutamine-independent ammonia release is subtracted, the ratio falls to 130 - 47 = 83/68or 1.22; furthermore, subtracting a similar ammonia blank from the ammonia released by acidotic kidneys, $297 - 47 = 250 \,\mu\text{moles}$, and dividing by glutamine removed, 154 µmoles, gives a ratio of 1.62. In previous work, employing dextran in place of albumin, I observed an ammonia/glutamine ratio of 1.4 in nonacidotic, increasing to 1.8 in acidotic rat kidneys (3, 4, 6). Subsequently, the role of a glutaminase- γ -glutamyltransferase was revealed in a series of studies (5, 6, 7, 15)culminating in the isolation of the enzyme from the soluble fraction (15).

The exact identity of this glutaminase- γ -

glutamyltransferase is at present unclear. It clearly is not γ -glutamyltranspeptidase (5, 15) and is probably not glutamine synthetase (although this enzyme complex is capable of glutaminase- γ -glutamyltransferase activity and is inhibited by MSO) since synthetase is predominantly microsomal (15, 16, 17) while the present activity is predominantly soluble (Table III, 15). Another enzyme, γ -glutamylcysteine synthetase, is a soluble protein (18) and inhibited by MSO (19), but does not utilize glutamine (20). Consequently, further studies are required to determine the exact identity of this activity.

Summary. Glutaminase- γ -glutamyltransferase contributes some 30% of the ammonia produced from glutamine by mildly acidotic rat kidneys. The enzyme is localized in the cytosol and its inhibition results in an ammonia produced per glutamine extracted ratio of 2.0. The results are therefore consistent with a dual glutamine-utilizing system, one cytoplasmic and the other mitochondrial, in the functioning rat kidney.

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 Damian, A. C., and Pitts, R. F., Amer. J. Physiol. 218, 1249 (1970).

- 2. Goldstein, L., Amer. J. Physiol. 213, 983 (1967).
- 3. Welbourne, T. C., Amer. J. Physiol. 226, 544 (1974).
- Phenix, P., and Welbourne, T. C., Amer. J. Physiol. 228, 1289 (1975).
- Wadoux, P., and Welbourne, T. C., Canad. J. Biochem. 53, 930 (1975).
- Welbourne, T. C., Proc. Soc. Exp. Biol. Med. 152, 64 (1976).
- 7. Welbourne, T. C., Med. Clin. N. Amer. 59, 629 (1975).
- 8. Bowman, R. H., J. Biol. Chem. 245, 1604 (1970).
- 9. Welbourne, T. C., Canad. J. Physiol. Pharmacol. 158, 883 (1972).
- 10. Chen, R. F., J. Biol. Chem. 242, 173 (1967).
- DeDuve, C., Pressman, B. C., Gianetto, R., Wattiaux, R., and Applermans, F., Biochem. J. 60, 604 (1955).
- Gornall, A. G., Bardawill, C. J., and Davis, M. M., J. Biol. Chem. 177, 751 (1949).
- 13. Hems, D. A., Biochem. J. 130, 671 (1972).
- 14. Ross, B., Clin. Sci. 50, 493 (1976).
- 15. Welbourne, T. C., "Biochemical Aspects of Kidney Function." Huber, Bern (1978).
- 16. Wu, C., Biochim. Biophys. Acta 77, 487 (1963).
- 17. Herzfeld, A., Biochem. J. 133, 49 (1973).
- Orlowski, M., and Meister, A., Biochemistry 10, 372 (1971).
- Richman, P. G., Orlowski, M., and Meister, A., J. Biol. Chem. 248, 6684 (1973).
- Sekura, R., and Meister, A., J. Biol. Chem. 252, 2599 (1977).

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